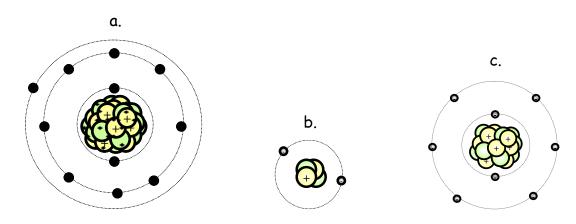
Name:	Section:
Understanding Periods and Groups	
	Using your Periodic Table of Elements and your knowledge of groups, answer the following questions.
1. How n	nany periods make up the periodic table?
2. How n	nany groups make up the periodic table?
3. How n six (6	nany elements make up period one (1)? Period three (3)? Period
	<b>,</b>
	nany elements make up group seven (7)? Group fourteen (14)? eighteen (18)?
	the element that resides here:  Group eighteen (18) - Period five (5)
b.	Period seven (7) - Group two (2)
C.	Group one (1) - Period one (1)
d.	Period three (3) - Group sixteen (16)
e.	Group ten (10) - Period six (6)
f.	Period one (1) - Group eight (8)

- 6. Which of the following pairs would most-likely have similar physical and chemical properties? Explain your answer.
  - a. Lithium and Selenium
  - b. Vanadium and Radon
  - c. Sodium and Potassium
- 7. Which of the following represent an atom from period three? Explain your answer.



8. Which of the following atoms could reside in the same group? Explain your answer.

